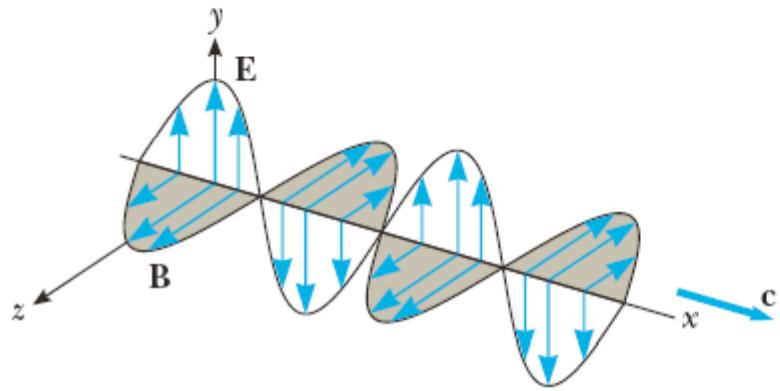


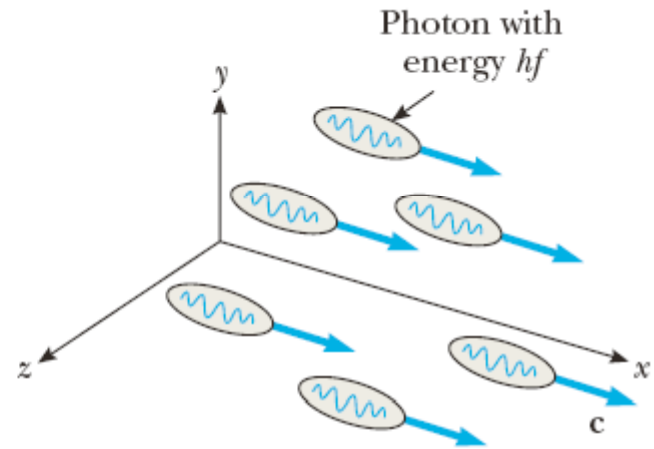


Physics 2D Lecture Slides  
Week of May 4, 2009  
Part 2

(Oleg Shpyrko)  
Sunil Sinha  
UCSD Physics



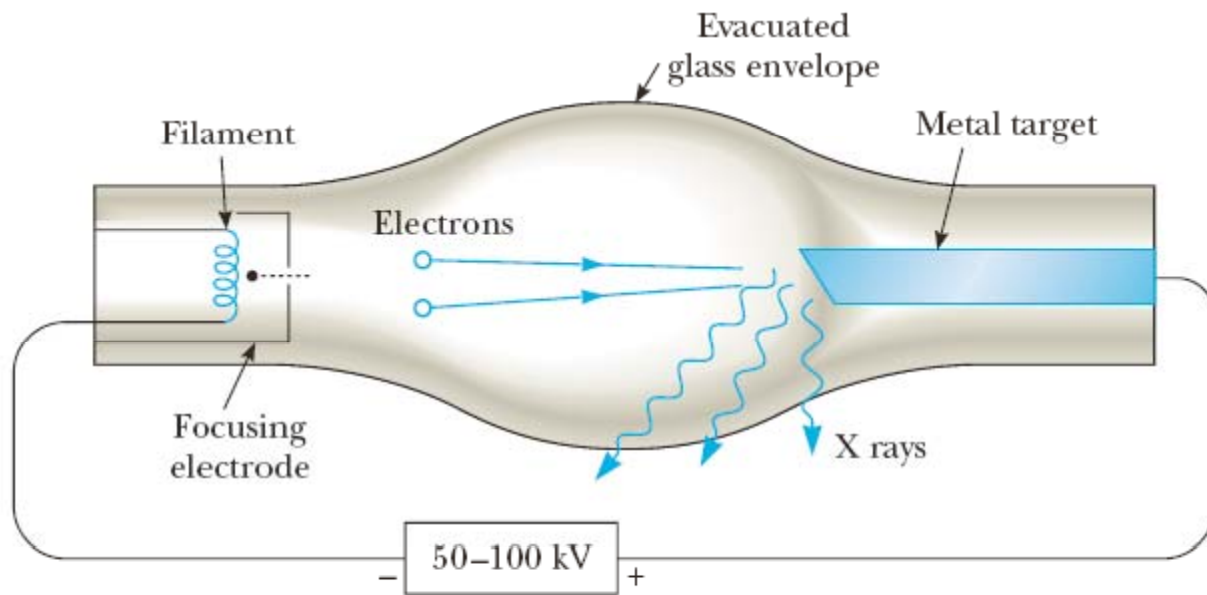
(a)

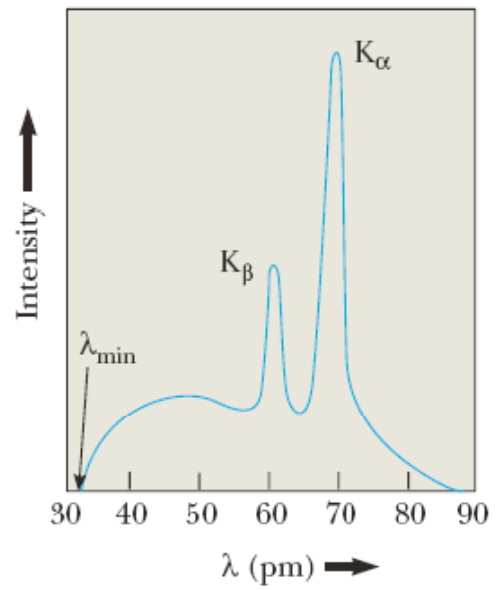
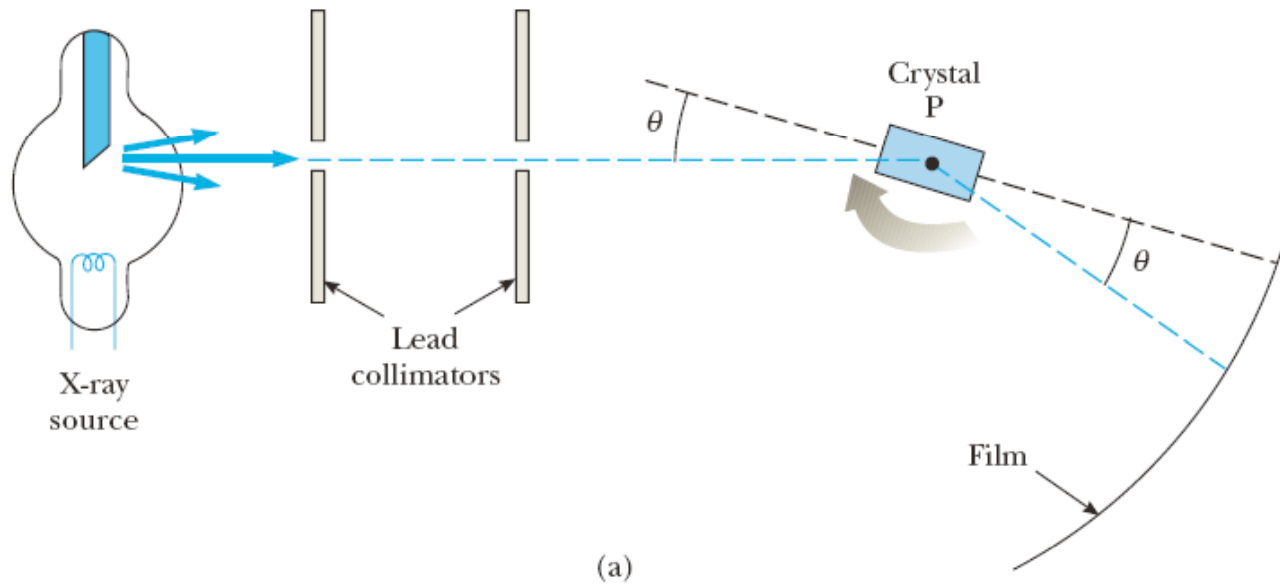


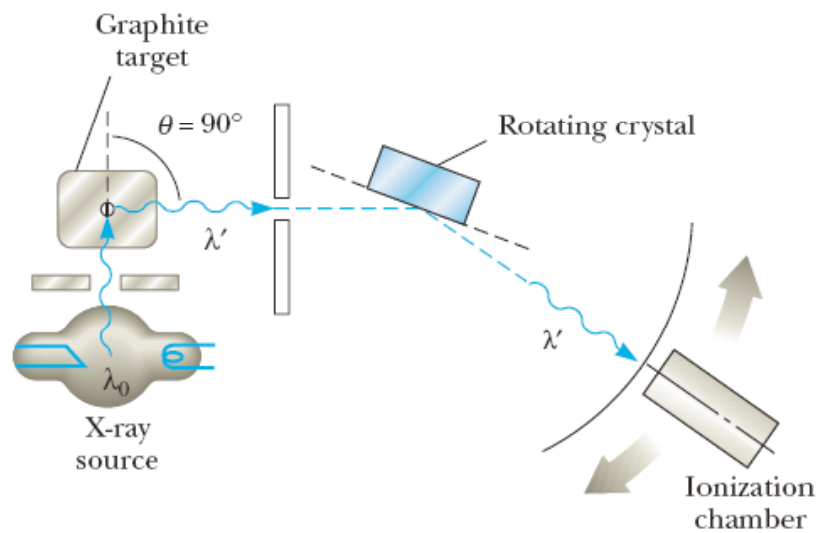
(b)

$$p = \frac{E}{c}$$

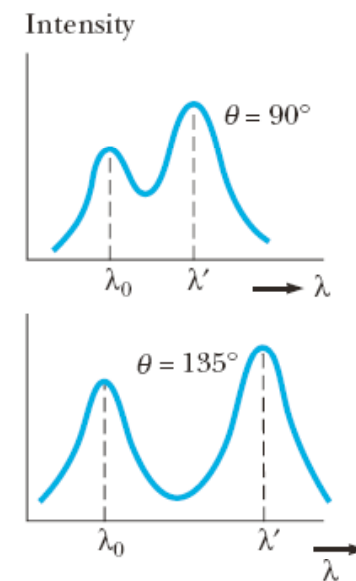
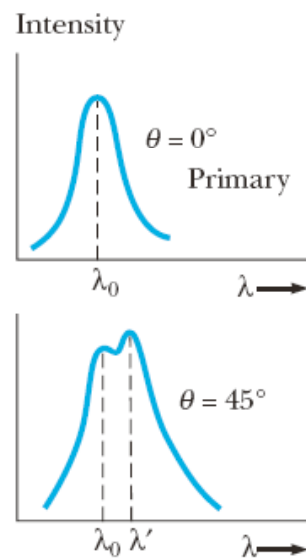
$$E^2 = (mc^2)^2 + (pc)^2$$







(a)



(b)



de Broglie's matter-wave theory

Louis de Broglie

# Bohr's Explanation of Hydrogen like atoms

- Bohr's Semiclassical theory explained some spectroscopic data → Nobel Prize : 1922
- The “hotch-potch” of classical & quantum attributes left many (Einstein) unconvinced
  - “appeared to me to be a miracle – and appears to me to be a miracle today ..... One ought to be ashamed of the successes of the theory”
- Problems with Bohr's theory:
  - Failed to predict INTENSITY of spectral lines
  - Limited success in predicting spectra of Multi-electron atoms (He)
  - Failed to provide “time evolution ” of system from some initial state
  - Overemphasized Particle nature of matter-could not explain the wave-particle duality of light
  - No general scheme applicable to non-periodic motion in subatomic systems
- Without fundamental insight ...raised the question : Why was Bohr successful?

## Prince Louis de Broglie & Matter Waves

- Key to Bohr atom was Angular momentum quantization
- Why this Quantization:  $mvr = |L| = nh/2\pi$  ?
- Invoking symmetry in nature, Louis de Broglie conjectured:

**Because photons have wave  
and particle like nature →  
particles may have wave like  
properties !!**

**Electrons have  
accompanying “pilot” wave  
(not EM) which guide  
particles thru spacetime**





## A PhD Thesis Fit For a Prince

- **Matter Wave !**
  - **“Pilot wave” of  $\lambda = h/p = h / (\gamma m v)$**
  - **frequency  $f = E/h$**
  
- **Consequence:**
  - **If matter has wave like properties then there would be interference (destructive & constructive)**
    - **Use analogy of standing waves on a plucked string to explain the quantization condition of Bohr orbits**

# Matter Waves : How big, how small

1. Wavelength of baseball,  $m=140\text{g}$ ,  $v=27\text{m/s}$

$$\lambda = \frac{h}{p} = \frac{h}{mv} = \frac{6.63 \times 10^{-34} \text{ J}\cdot\text{s}}{(.14\text{kg})(27\text{m/s})} = 1.75 \times 10^{-34} \text{ m}$$

$\Rightarrow$   $\lambda_{\text{baseball}} \ll \ll$  size of nucleus

$\Rightarrow$  Baseball "looks" like a particle

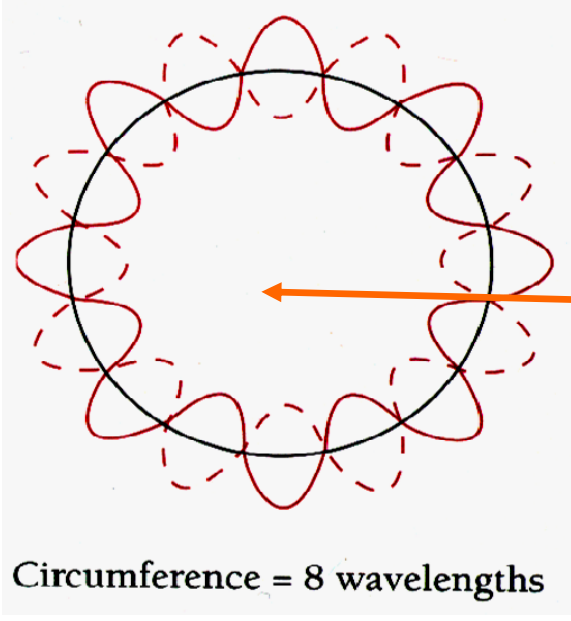
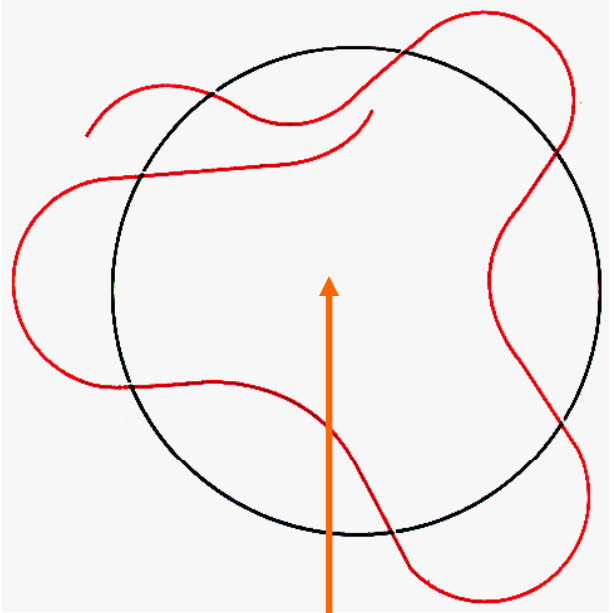
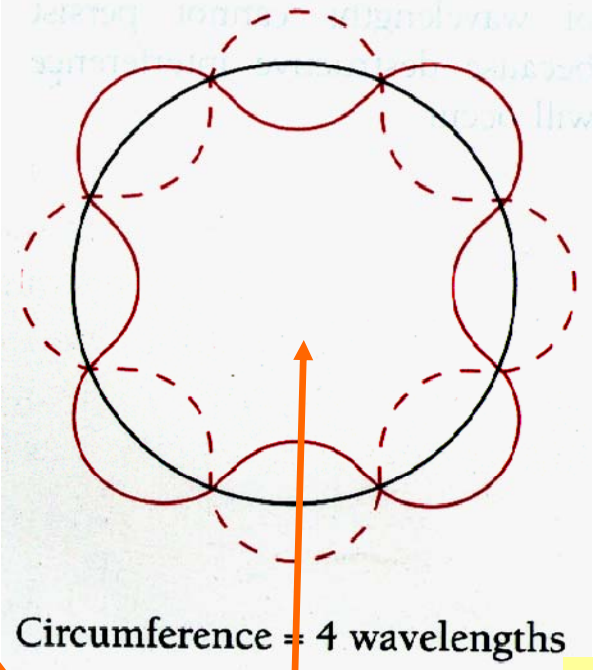
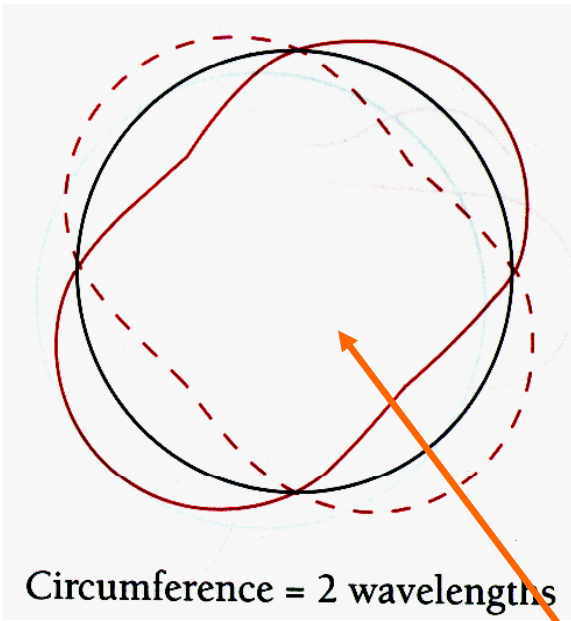
2. Wavelength of electron  $K=120\text{eV}$  (assume NR)

$$\begin{aligned} K &= \frac{p^2}{2m} \Rightarrow p = \sqrt{2mK} \\ &= \sqrt{2(9.11 \times 10^{-31})(120\text{eV})(1.6 \times 10^{-19})} \\ &= 5.91 \times 10^{-24} \text{ Kg}\cdot\text{m/s} \end{aligned}$$

$$\lambda_e = \frac{h}{p} = \frac{6.63 \times 10^{-34} \text{ J}\cdot\text{s}}{5.91 \times 10^{-24} \text{ kg}\cdot\text{m/s}} = 1.12 \times 10^{-10} \text{ m}$$

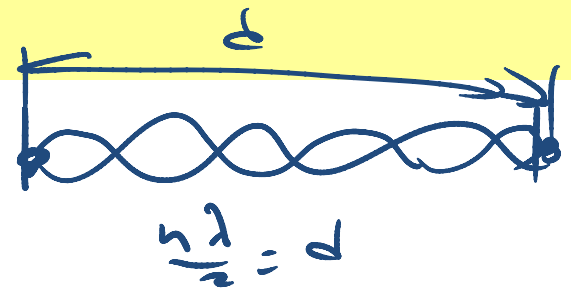
$\Rightarrow$   $\lambda_e \approx$  Size of atom !!

# Models of Vibrations on a Loop: Model of e in atom



Modes of vibration when an integral # of  $\lambda$  fit into loop  
 (Standing waves) vibrations continue indefinitely

Fractional # of waves in a loop can not persist due to destructive interference



# De Broglie's Explanation of Bohr's Quantization

Standing waves in H atom:

Constructive interference when  
 $n\lambda = 2\pi r$

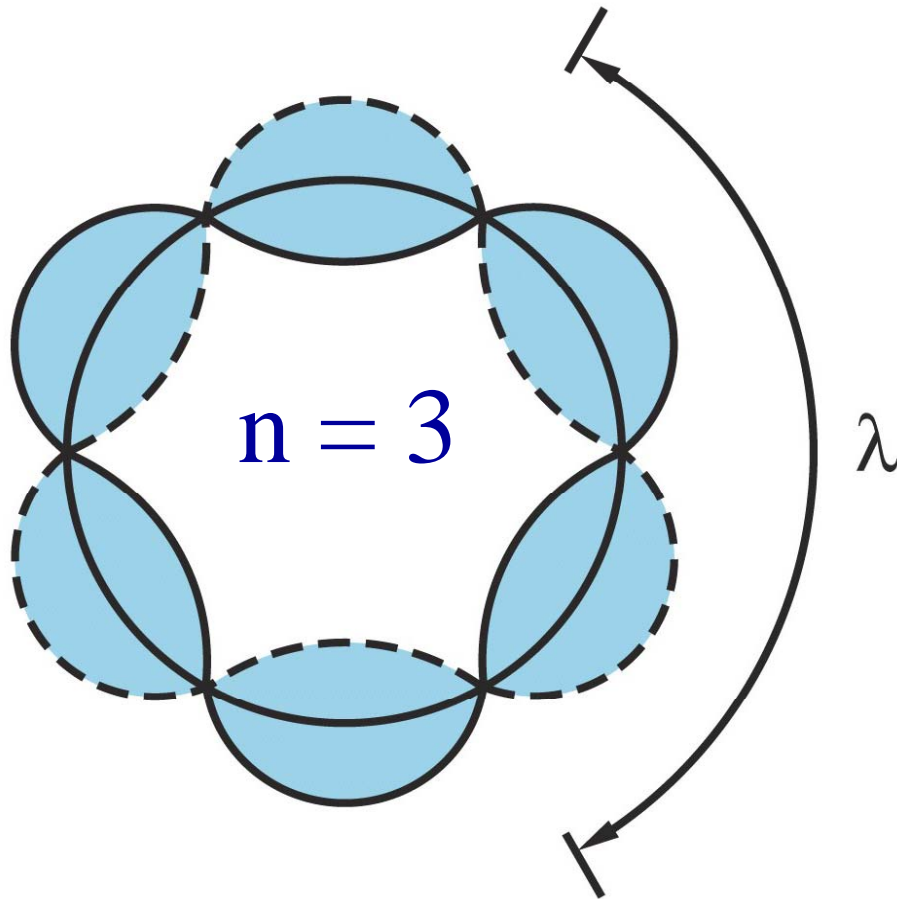
$$\text{since } \lambda = \frac{h}{p} = \frac{h}{mv} \quad \dots\dots(NR)$$

$$\Rightarrow \frac{nh}{mv} = 2\pi r$$

$$\Rightarrow \boxed{n\hbar = mvr}$$

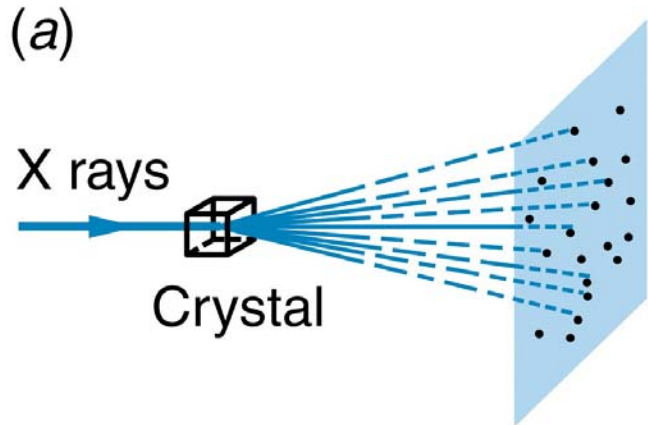
Angular momentum

Quantization condition!

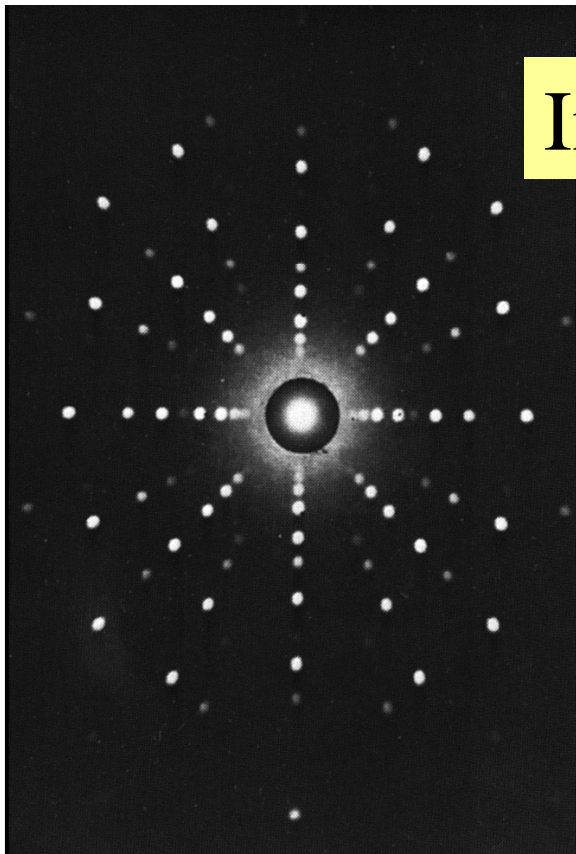


This is too intense ! Must verify such “loony tunes” with experiment

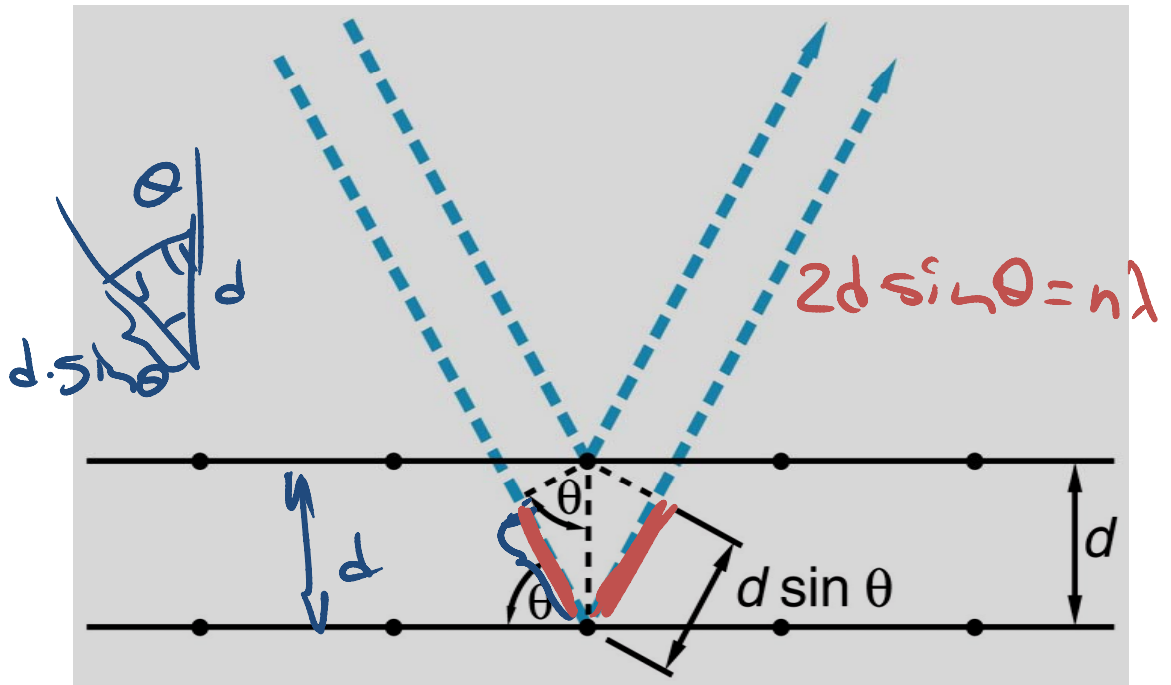
# Reminder: Light as a Wave : Bragg Scattering Expt

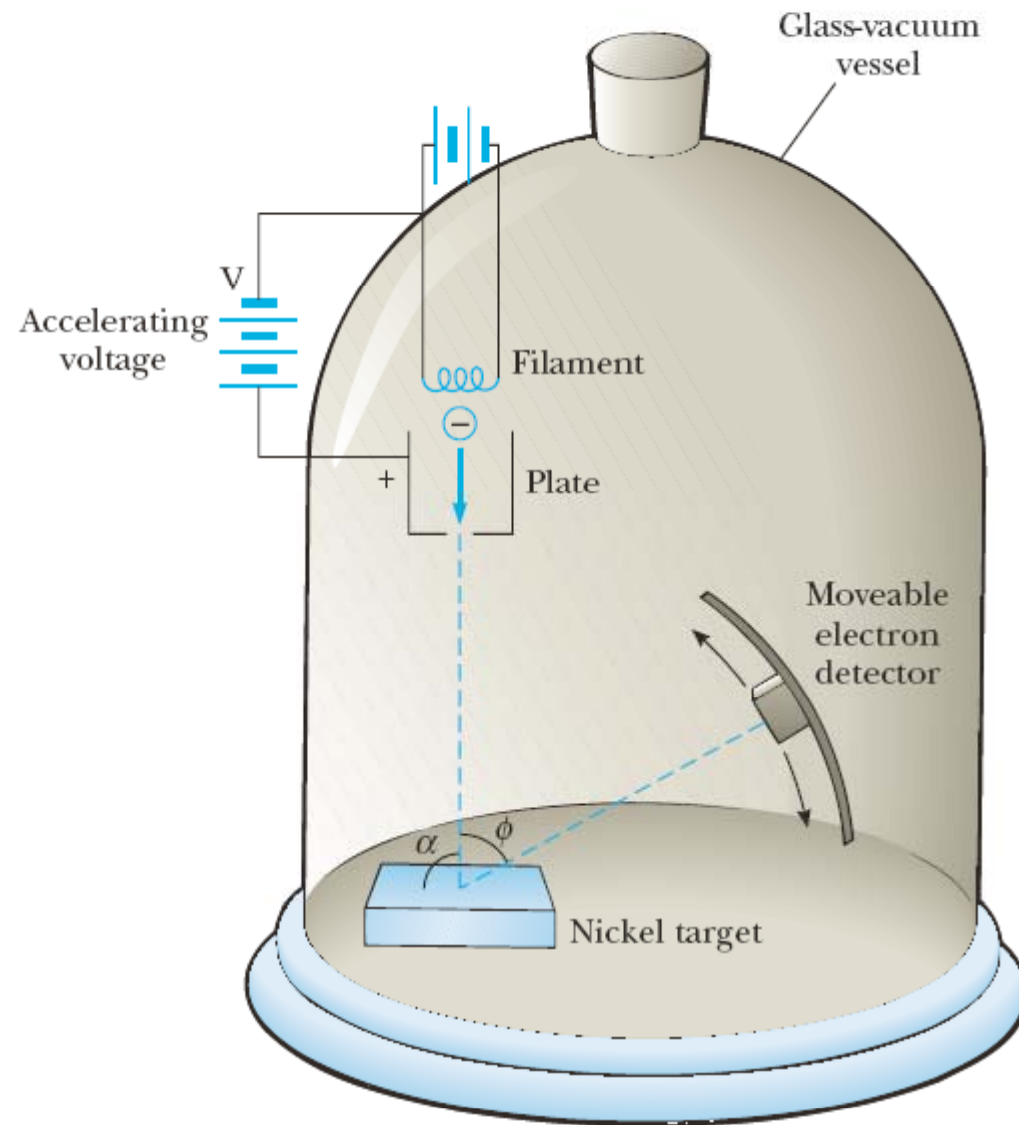


Range of X-ray wavelengths scatter  
Off a crystal sample  
X-rays constructively interfere from  
Certain planes producing bright spots



Interference  $\rightarrow$  Path diff =  $2d \sin \theta = n\lambda$



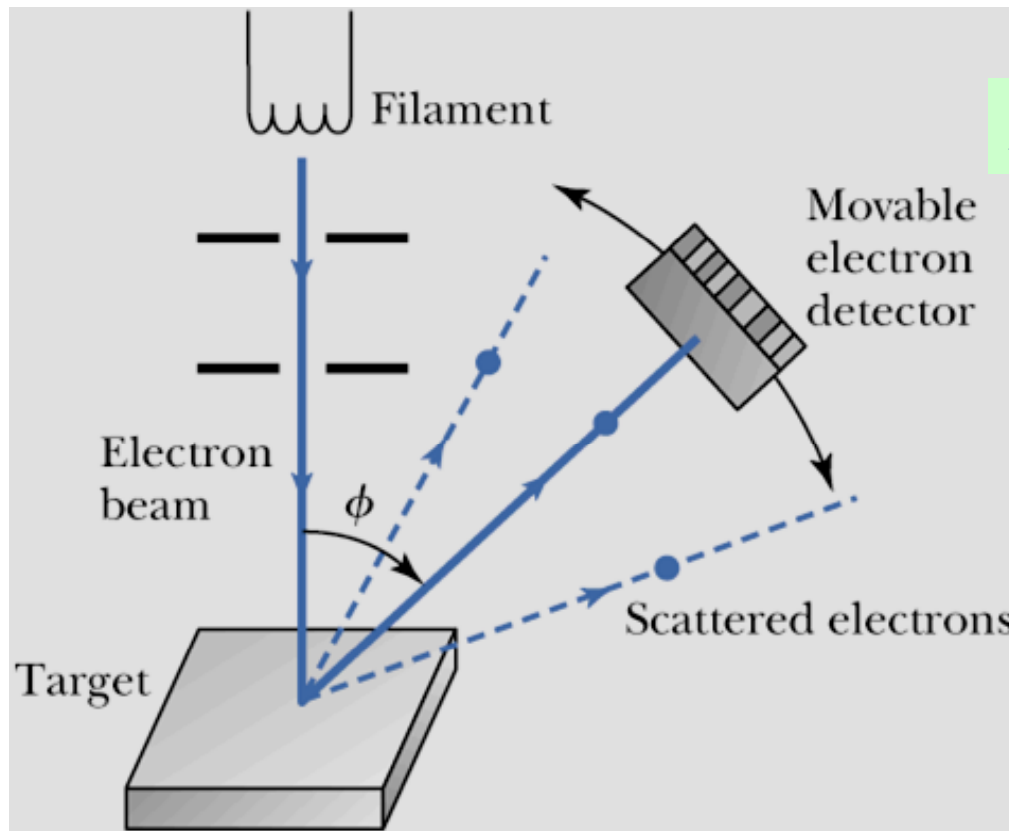


**Figure 5.4** A schematic diagram of the Davisson–Germer apparatus.

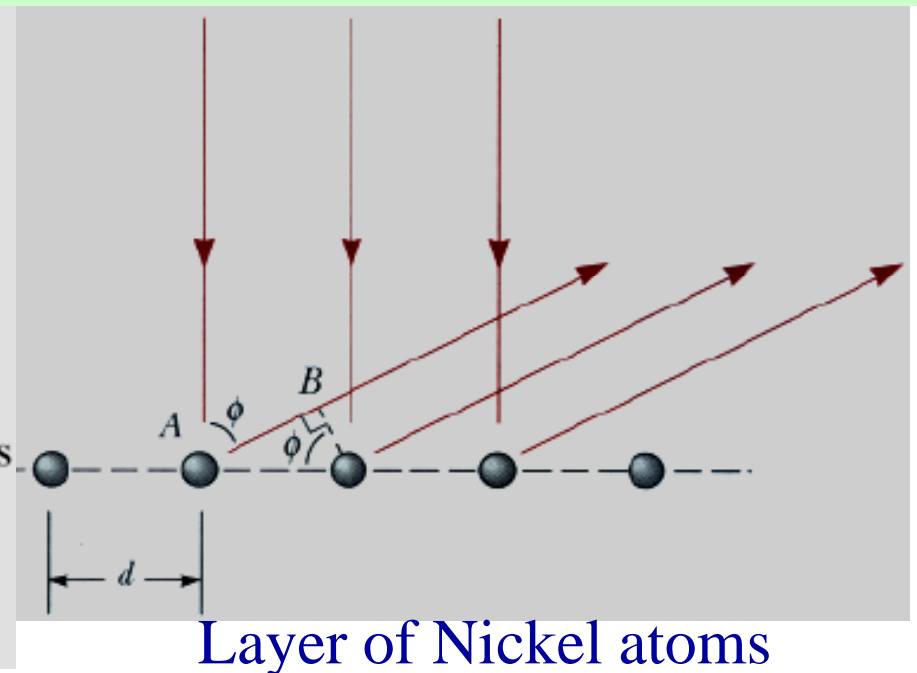
# Verification of Matter Waves: Davisson & Germer Expt

If electrons have associated wave like properties  $\rightarrow$  expect interference pattern when incident on a layer of atoms (reflection diffraction grating) with inter-atomic separation  $d$  such that

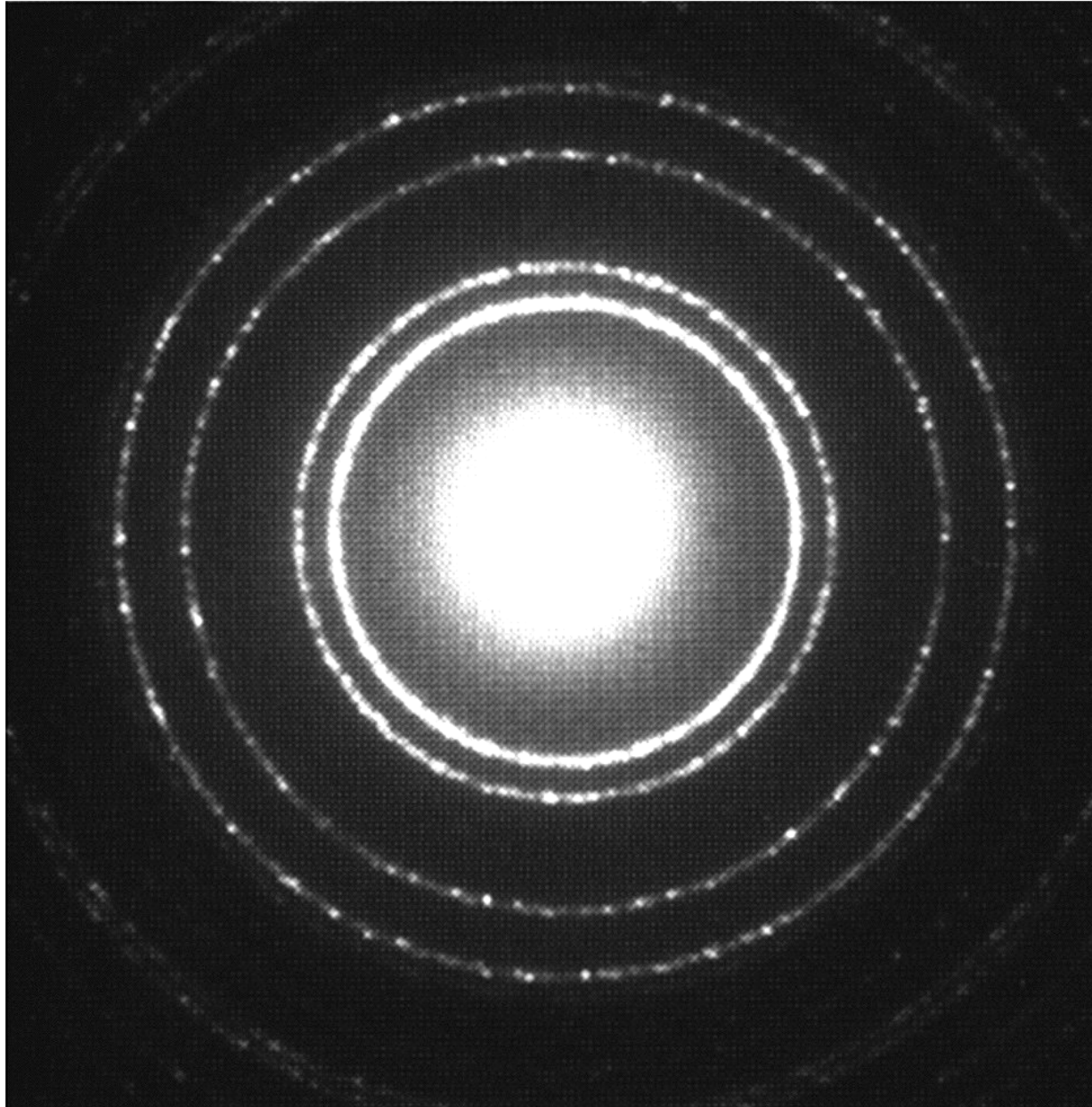
$$\text{path diff AB} = d \sin \phi = n \lambda$$



Atomic lattice as diffraction grating



## Electrons Diffract in Crystal, just like X-rays



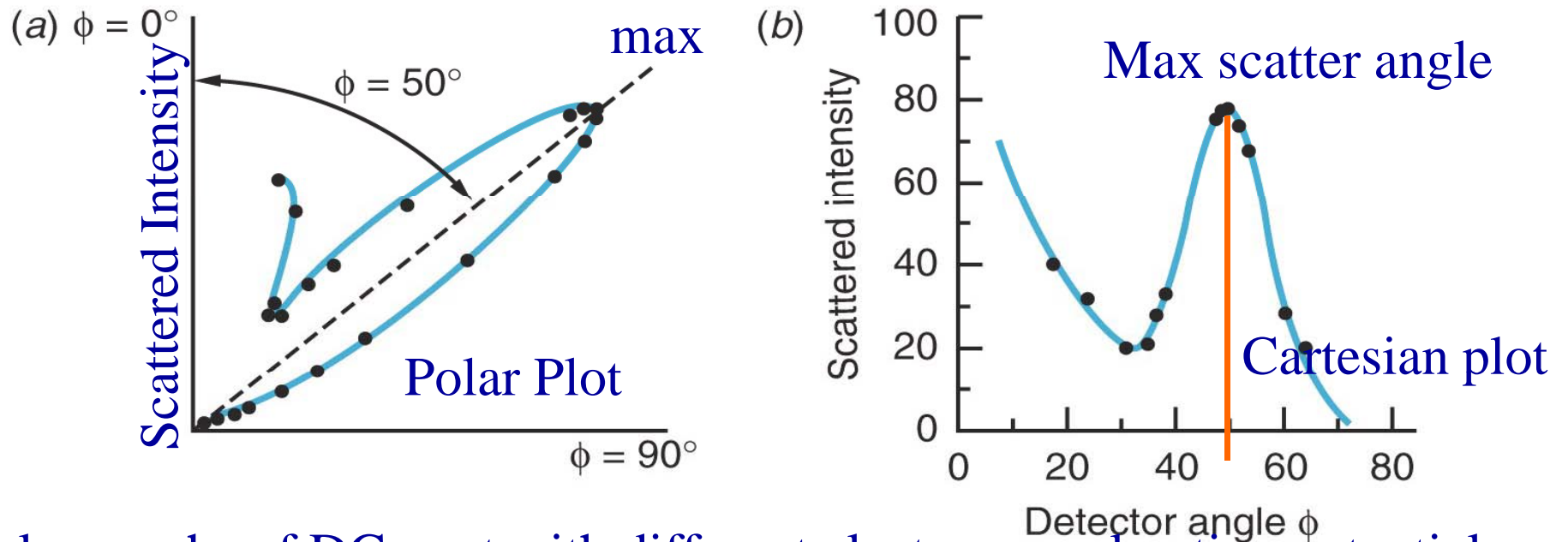
Diffraction pattern produced by 600eV electrons incident on a Al foil target

Notice the waxing and waning of scattered electron Intensity.

What to expect if electron had no wave like attribute

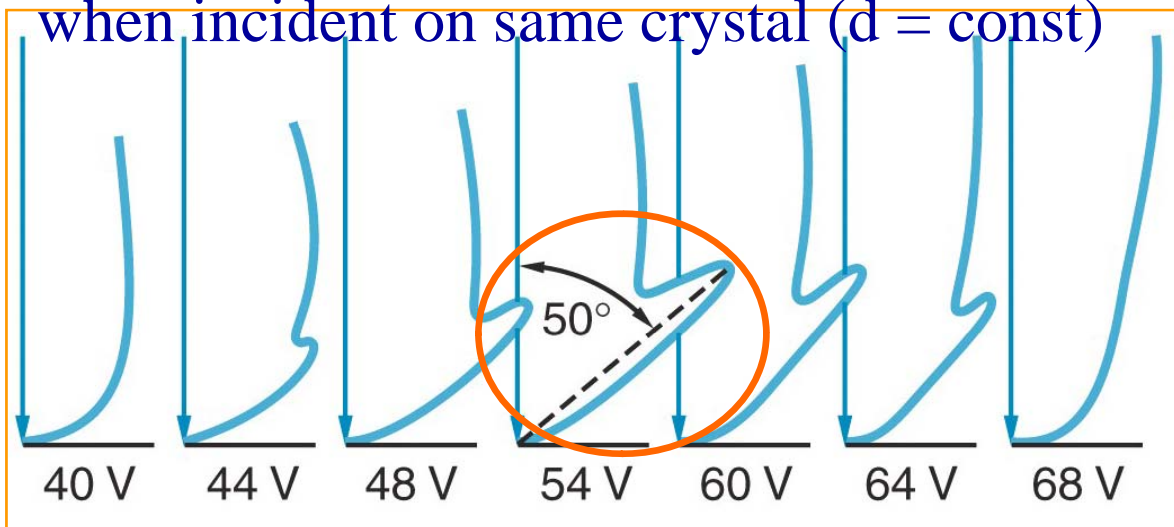


# Davisson-Germer Experiment: 54 eV electron Beam



Polar graphs of DG expt with different electron accelerating potential

when incident on same crystal ( $d = \text{const}$ )



Peak at  $\Phi = 50^\circ$   
when  $V_{\text{acc}} = 54 \text{ V}$

# Analyzing Davisson-Germer Expt with de Broglie idea

de Broglie  $\lambda$  for electron accelerated thru  $V_{\text{acc}}=54\text{V}$

$$\bullet \frac{1}{2}mv^2 = K = \frac{p^2}{2m} = eV \Rightarrow v = \sqrt{\frac{2eV}{m}} \quad ; \quad p = mv = m\sqrt{\frac{2eV}{m}}$$

If you believe de Broglie

$$\lambda = \frac{h}{p} = \frac{h}{mv} = \frac{h}{m\sqrt{\frac{2eV}{m}}} = \boxed{\frac{h}{\sqrt{2meV}} = \lambda^{\text{predict}}}$$

For  $V_{\text{acc}} = 54 \text{ Volts} \Rightarrow \lambda = 1.67 \times 10^{-10} \text{ m}$  (de Broglie)

Exptal data from Davisson-Germer Observation:

$d_{\text{nickel}} = 2.15 \text{ \AA} = 2.15 \times 10^{-10} \text{ m}$  (from Bragg Scattering)

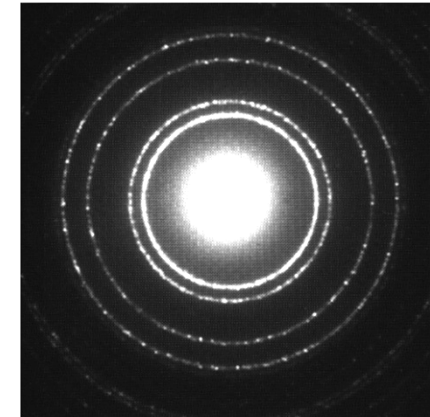
$\theta_{\text{diff}}^{\text{max}} = 50^\circ$  (observation from scattering intensity plot)

$$\boxed{\text{Diffraction Rule : } d \sin \phi = n\lambda}$$

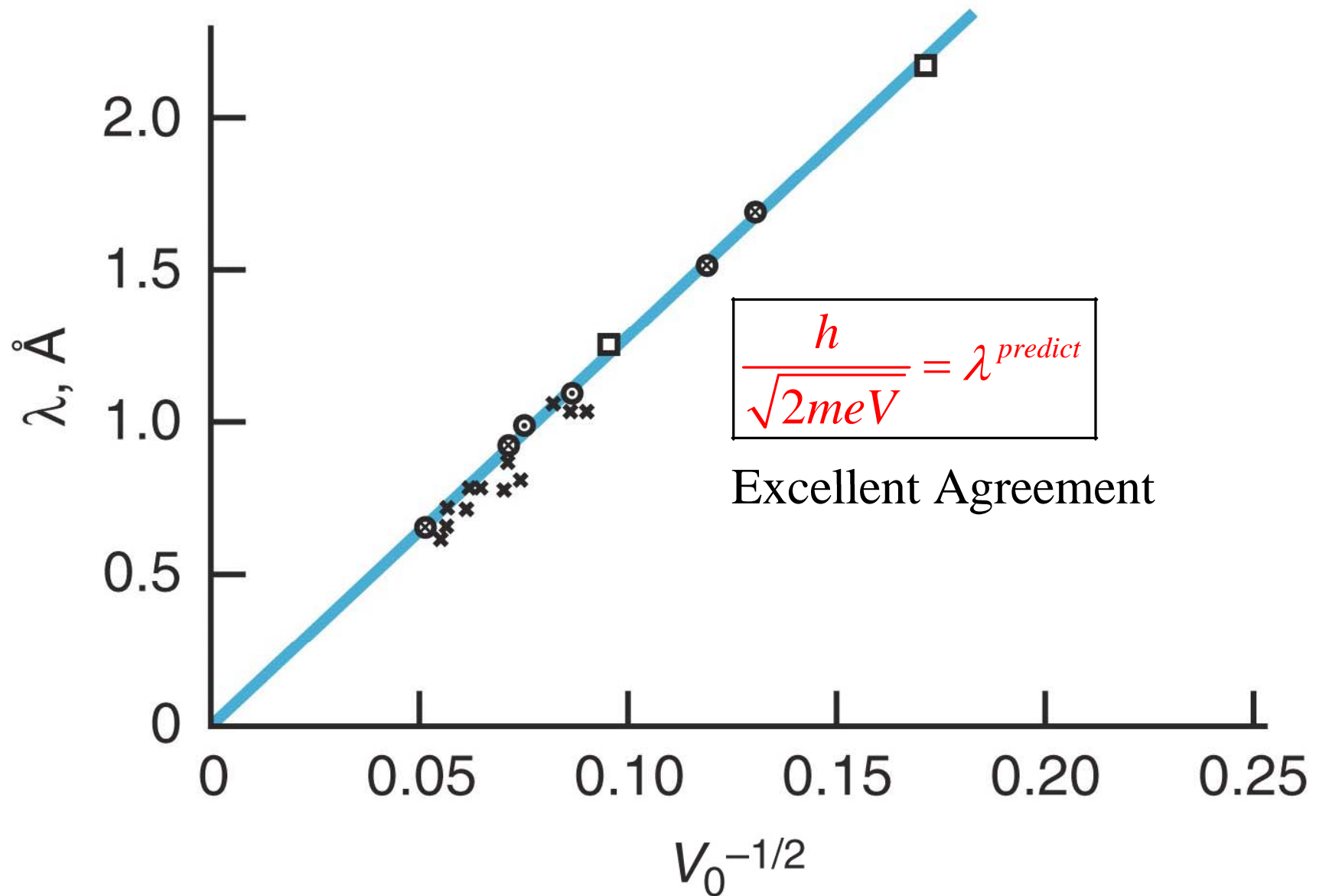
$\Rightarrow$  For Principal Maxima ( $n=1$ );  $\lambda^{\text{meas}} = (2.15 \text{ \AA})(\sin 50^\circ)$

$$\boxed{\begin{array}{l} \lambda^{\text{predict}} = 1.67 \text{ \AA} \\ \lambda^{\text{observ}} = 1.65 \text{ \AA} \end{array}}$$

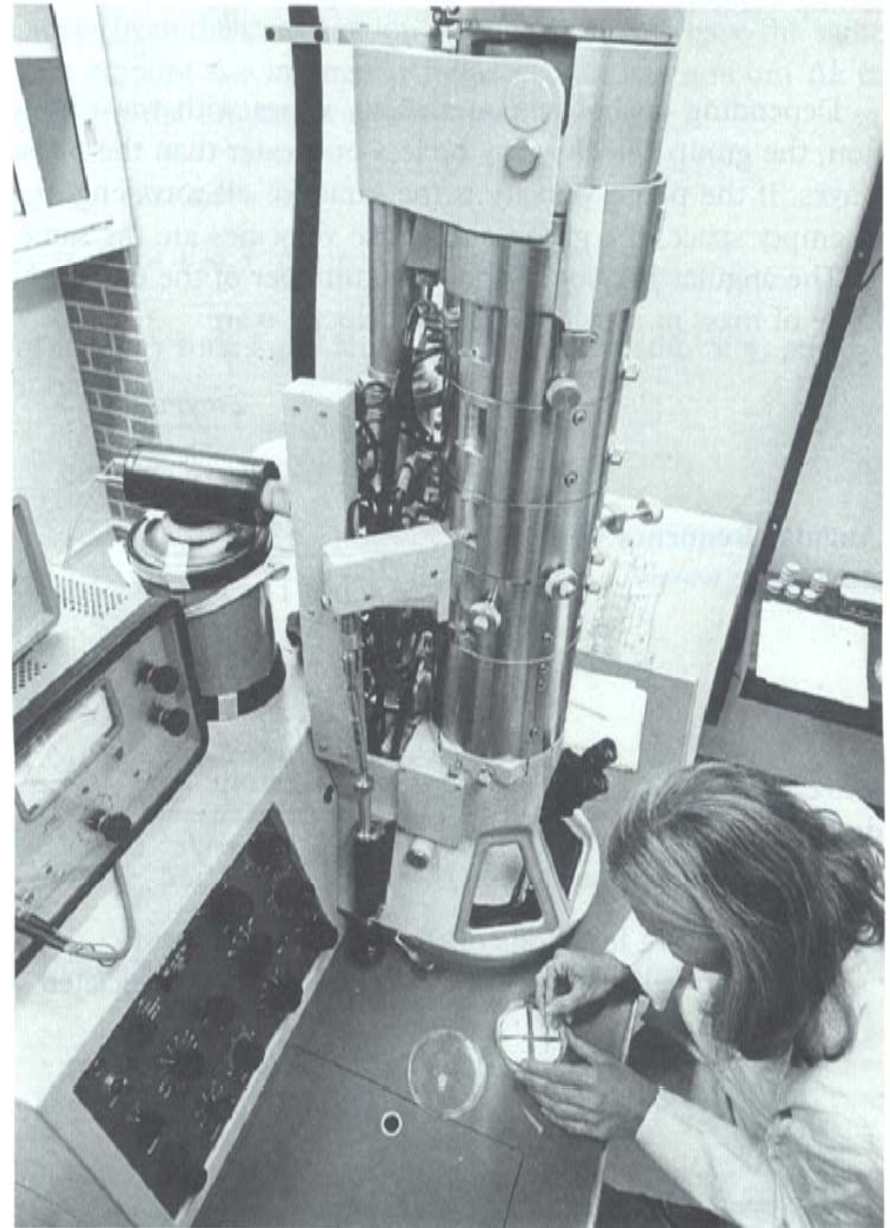
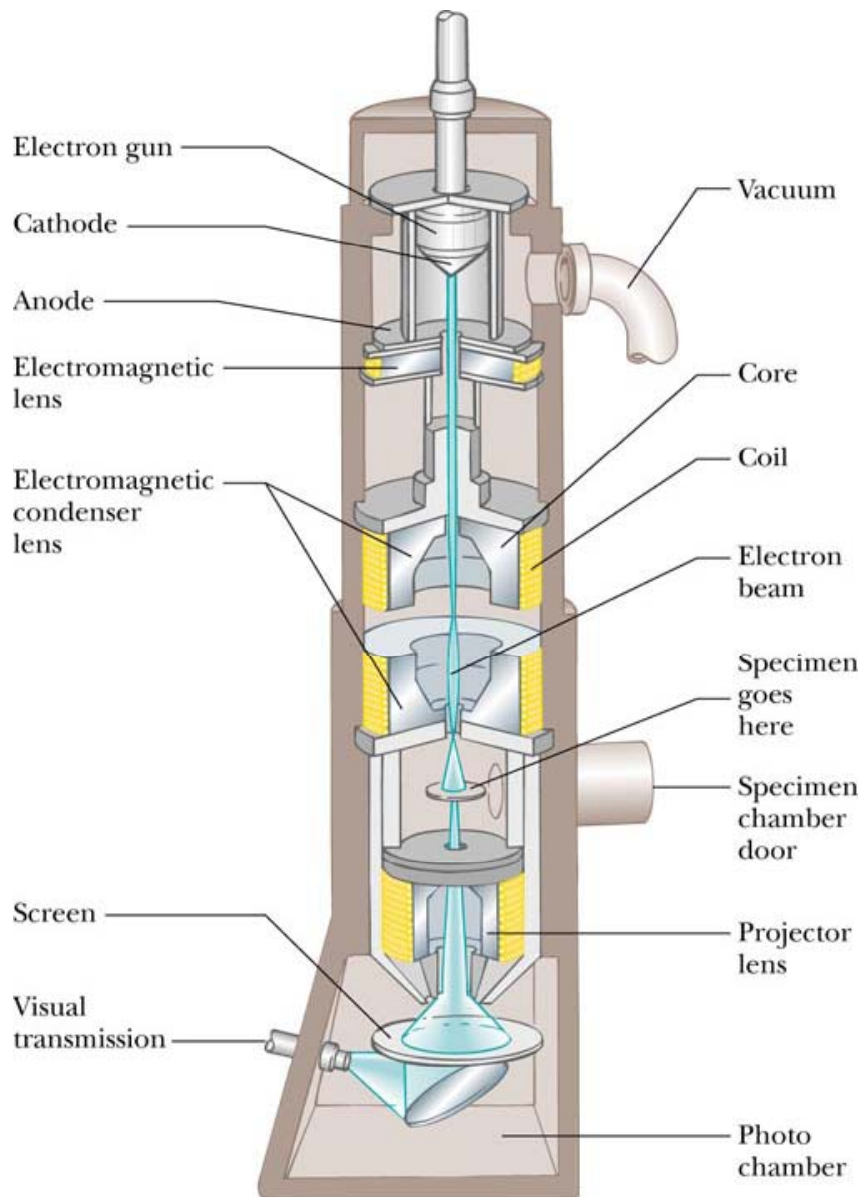
*Excellent agreement*



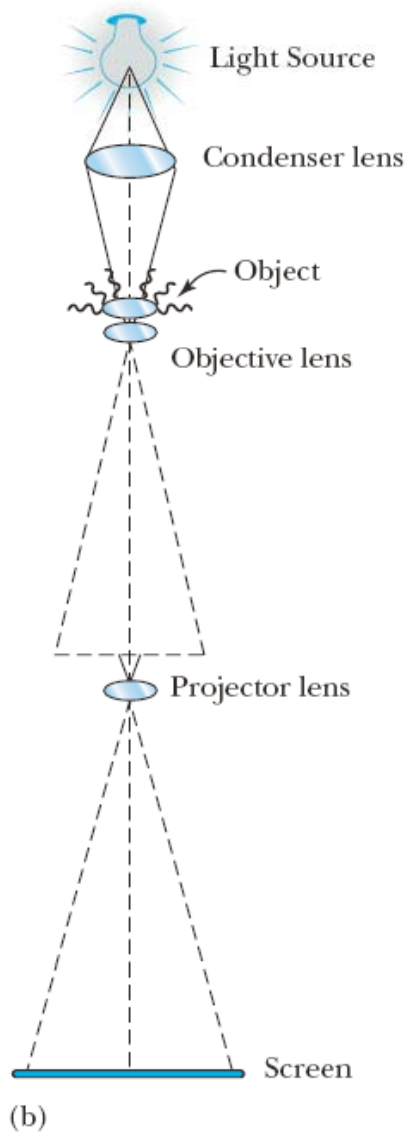
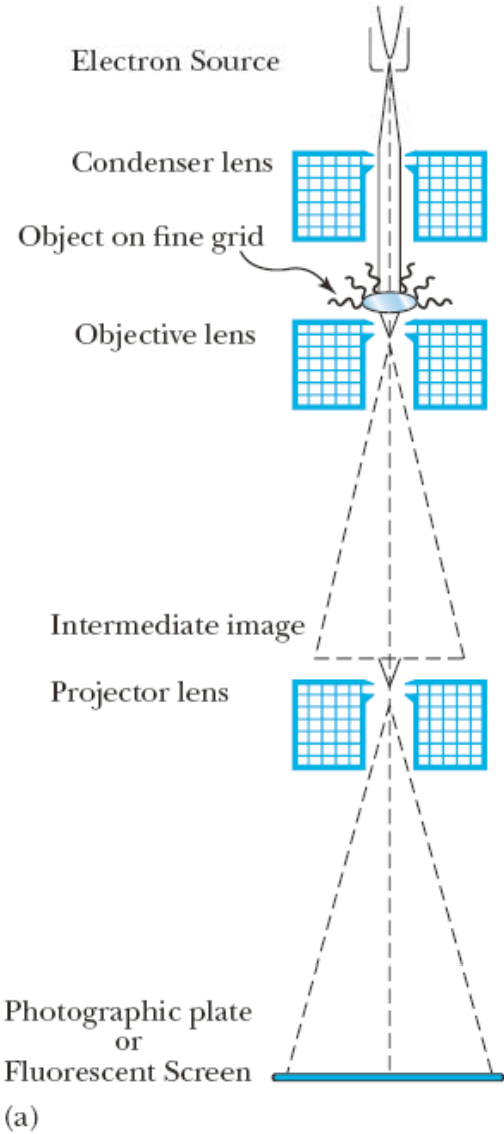
# Davisson Germer Experiment: Matter Waves !



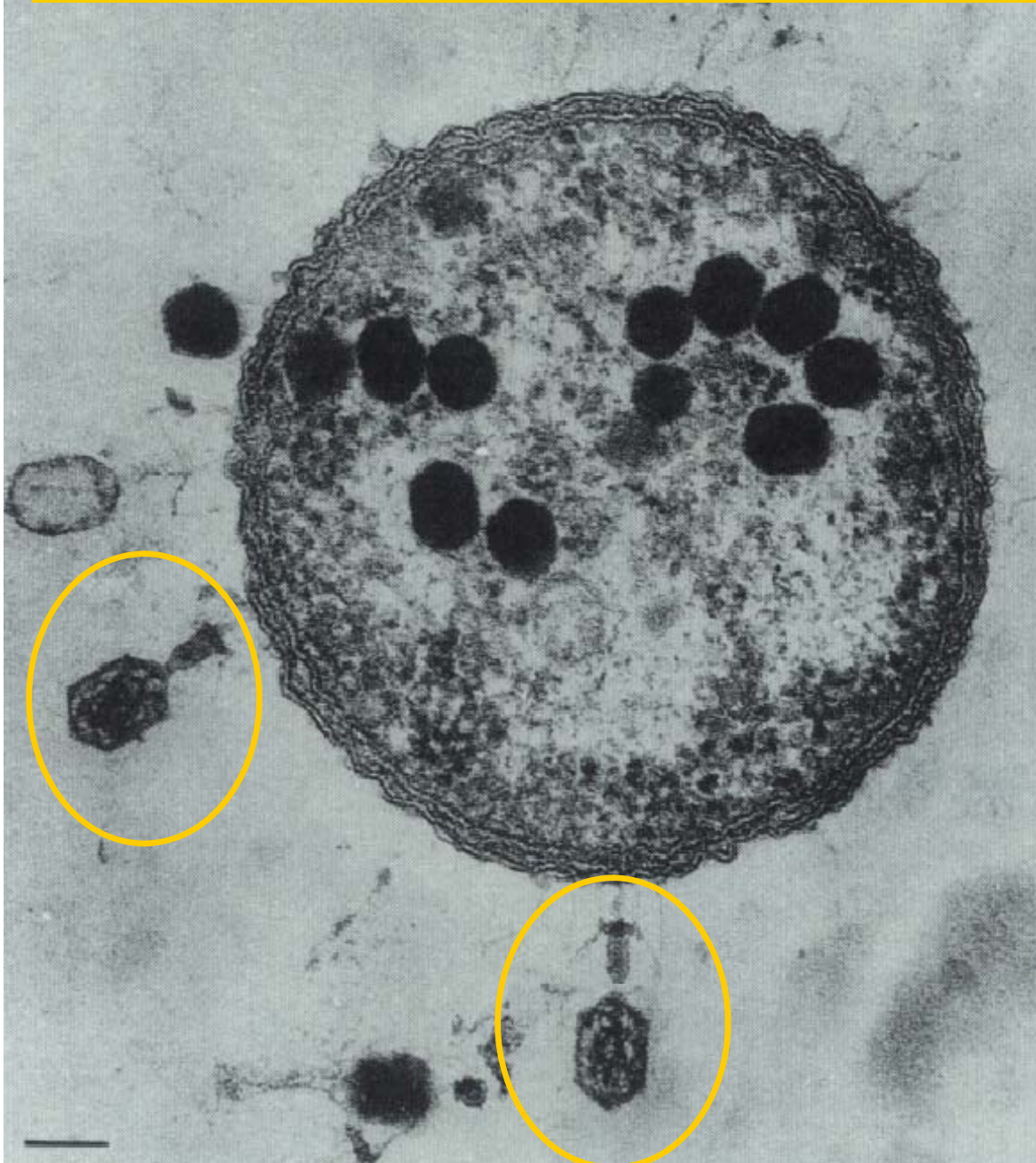
# Practical Application : Electron Microscope



# Electron Microscopy



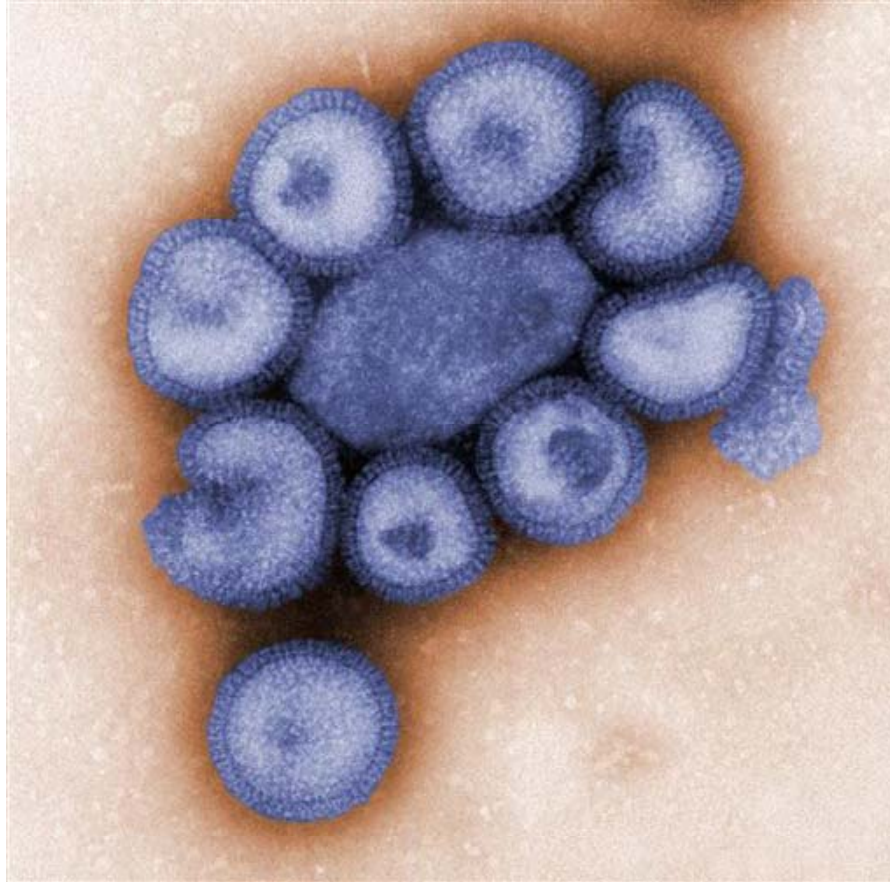
# Electron Microscope : Excellent Resolving Power



Electron Micrograph  
Showing Bacteriophage  
Viruses in E. Coli bacterium

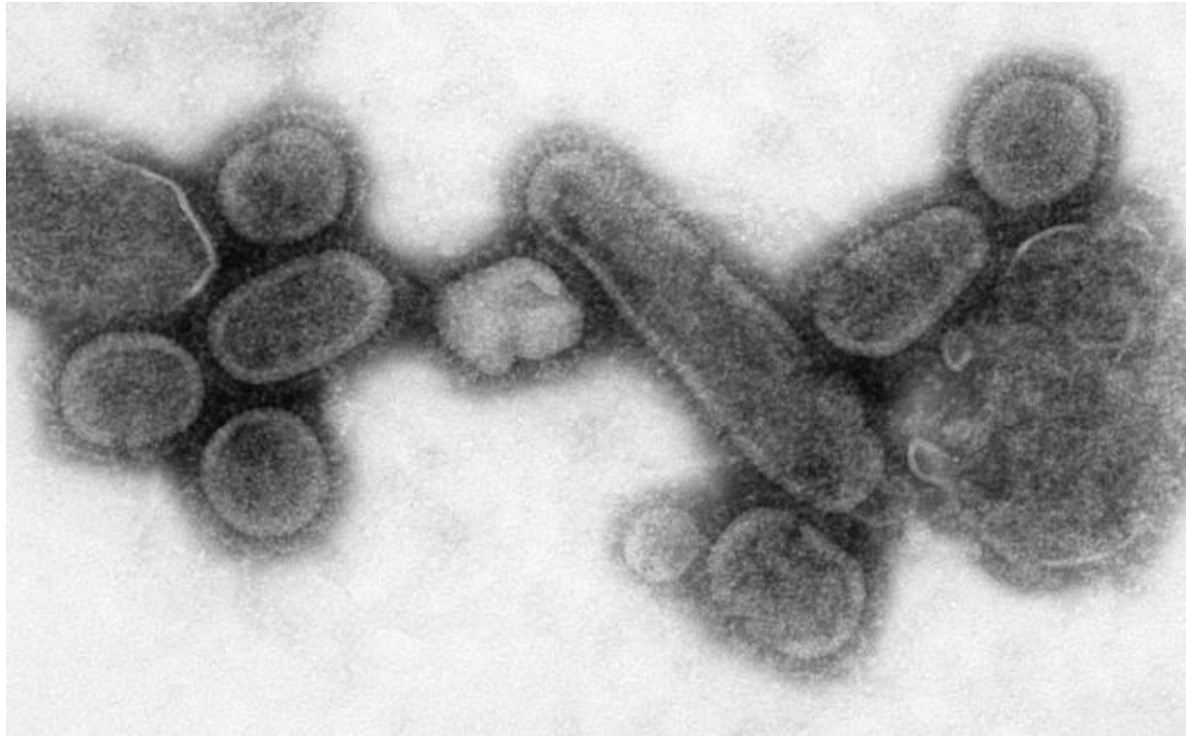
The bacterium is  $\cong 1\mu$  size

# Swine Flu Virus



Electron microscope image of the H1N1 virus, April 27, 2009, at the U.S. Centers for Disease Control and Prevention's headquarters in Atlanta, Georgia (AP Photo/Center for Disease Control and Prevention, C. S. Goldsmith and A. Balish) The viruses are 80–120 nanometres in diameter.

# 1918 Swine Flu Virus

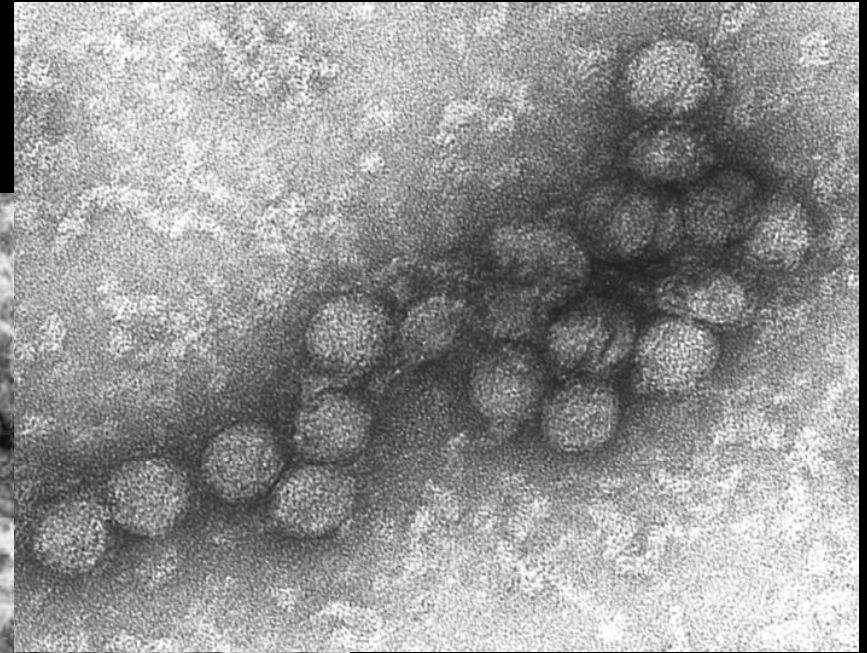
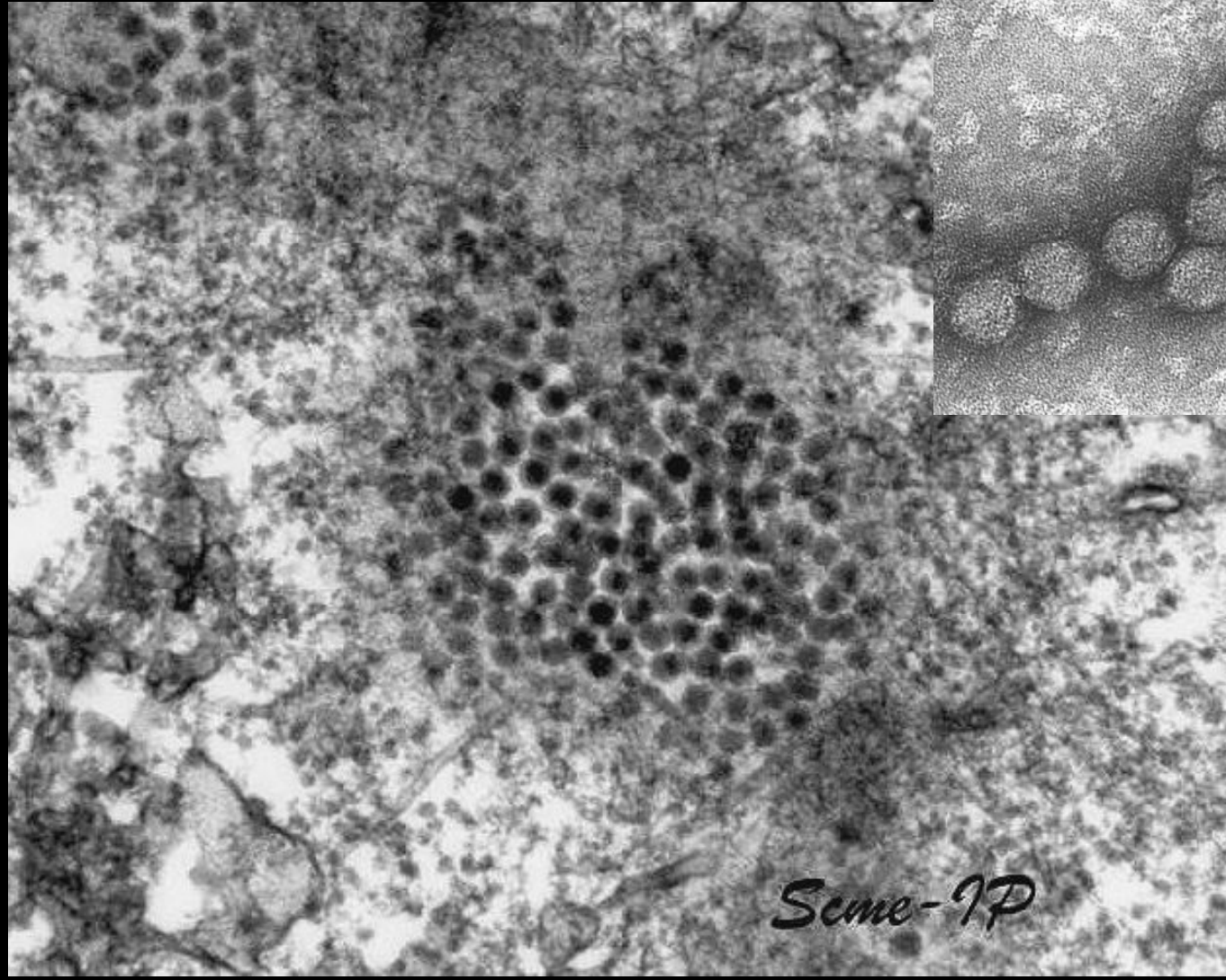


Negative stained transmission electron micrograph (TEM) showed recreated 1918 influenza virions that were collected from the supernatant of a 1918-infected Madin-Darby Canine Kidney (MDCK) cell culture 18 hours after infection.

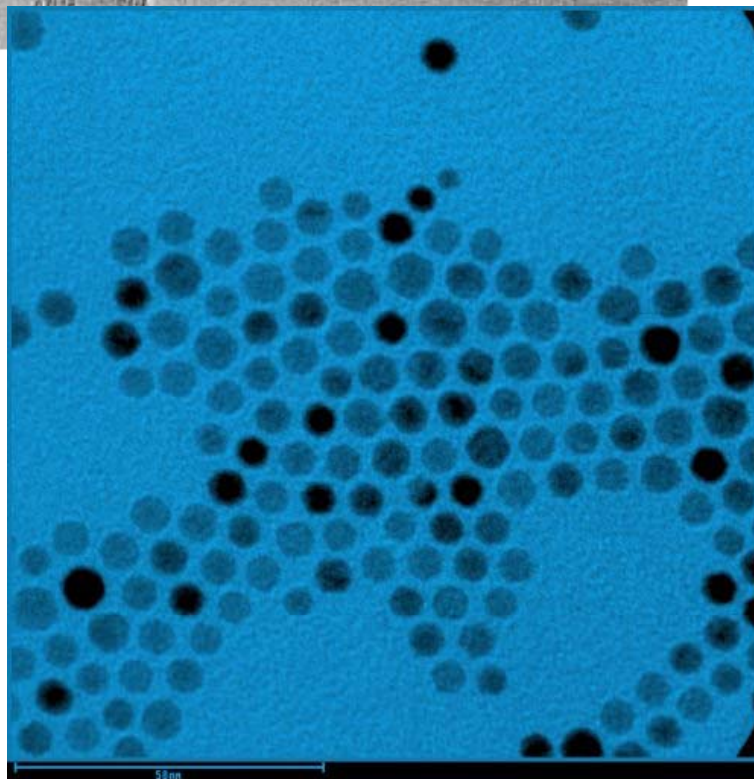
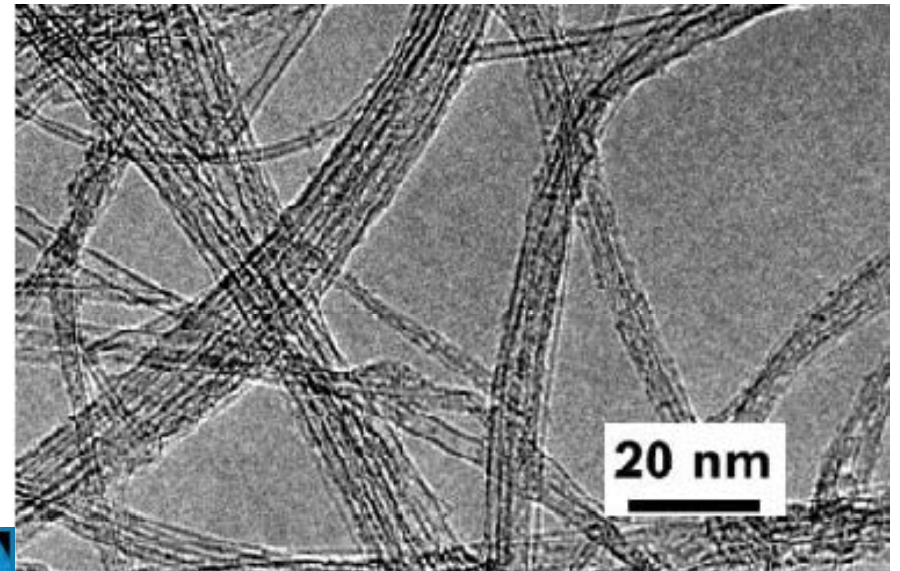
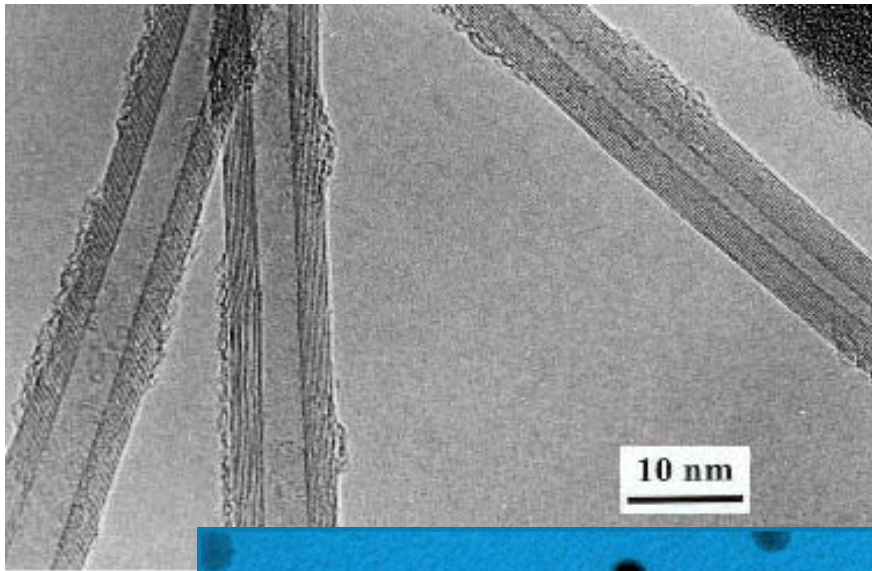
The 1918 Spanish flu epidemic was caused by an influenza A (H1N1) virus, killing more than 500,000 people in the United States, and up to 50 million worldwide.



# West Nile Virus extracted from a crow brain



## TEM pictures of Carbon Nanotubes



TEM of 6 nm Au Nanoparticles  
(Shpyrko Group, UCSD)

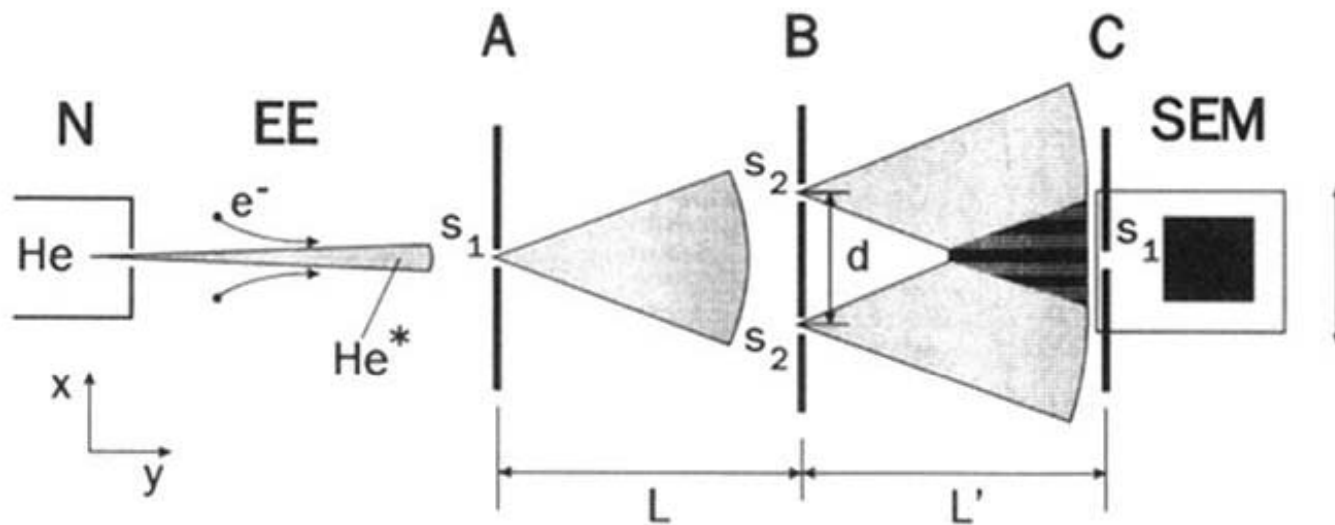
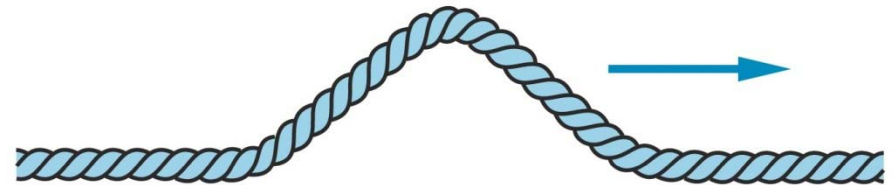


FIG. 2. Schematic representation of the experimental setup: nozzle system and gas reservoir N; electron impact excitation EE; entrance slit A, double slit B, and detector screen C; secondary electron multiplier SEM (mounted together with C on a translation stage). Dimensions:  $d = 8 \mu\text{m}$ ,  $L = L' = 64 \text{ cm}$ ; slit widths:  $s_1 = 2 \mu\text{m}$ ,  $s_2 = 1 \mu\text{m}$ .

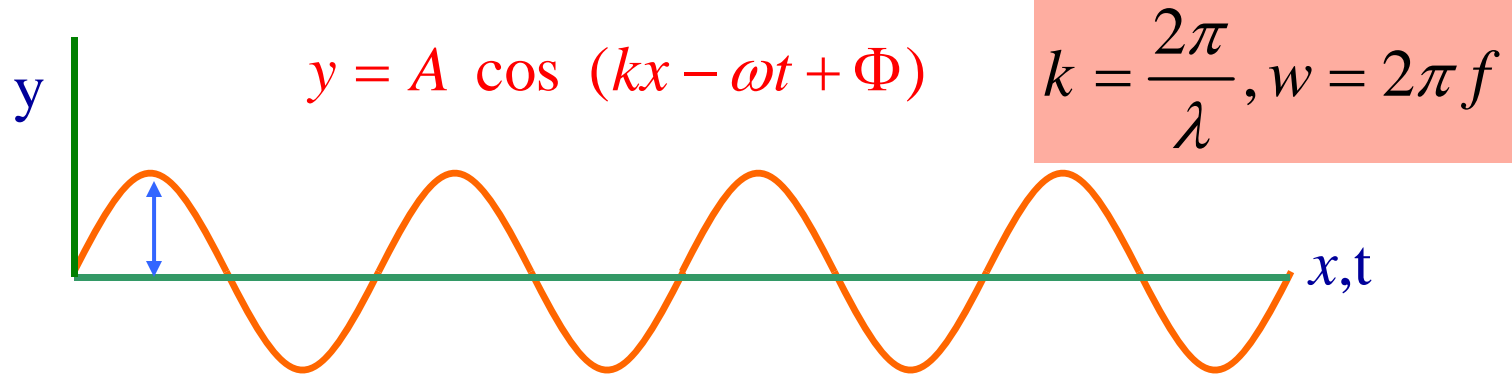
# Just What is Waving in Matter Waves ?

- For waves in an ocean, it's the water that "waves"
- For sound waves, it's the molecules in medium
- For light it's the **E & B** vectors
- What's waving for matter waves ?
  - It's the **PROBABILITY OF FINDING THE PARTICLE** that waves !
  - Particle can be represented by a wave packet in
    - Space
    - Time
    - Made by superposition of many sinusoidal waves of different  $\lambda$
    - It's a "pulse" of probability

Imagine Wave pulse moving along a string: its localized in time and space (unlike a pure harmonic wave)



# What Wave Does Not Describe a Particle



- What wave form can be associated with particle's pilot wave?
- A traveling sinusoidal wave?  $y = A \cos(kx - \omega t + \Phi)$
- Since de Broglie "pilot wave" represents particle, it must travel with same speed as particle .....(like me and my shadow)

Phase velocity ( $v_p$ ) of sinusoidal wave:  $v_p = \lambda f$

In Matter:

$$(a) \lambda = \frac{h}{p} = \frac{h}{\gamma m v}$$

$$(b) f = \frac{E}{h} = \frac{\gamma m c^2}{h}$$

$$\Rightarrow v_p = \lambda f = \frac{E}{p} = \frac{\gamma m c^2}{\gamma m v} = \frac{c^2}{v} > c!$$

Conflicts with  
 Relativity →  
 Unphysical

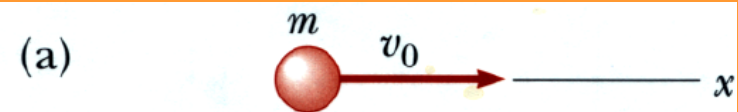
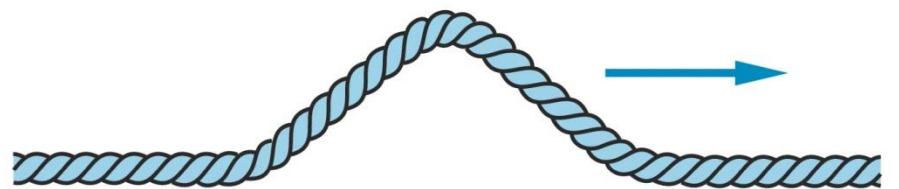
Single sinusoidal wave of infinite extent does not represent particle localized in space

Need "wave packets" localized  
 Spatially (x) and Temporally (t)

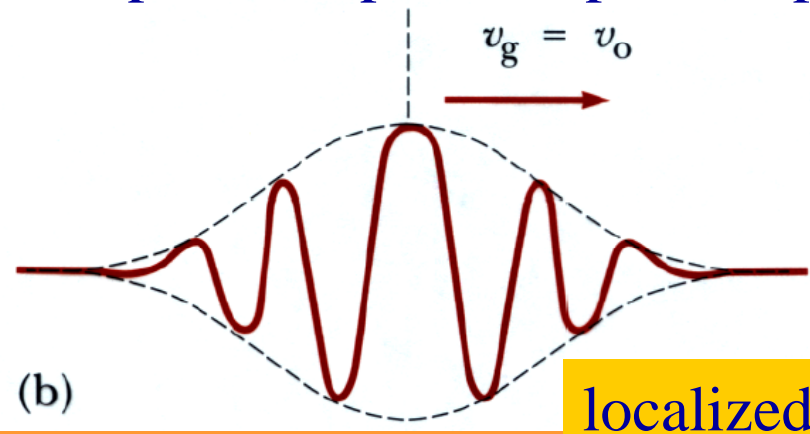
# Wave Group or Wave Pulse

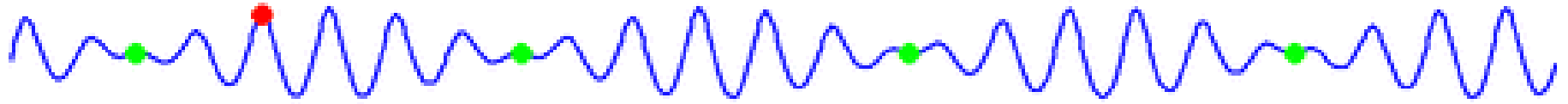
- Wave Group/packet:
  - Superposition of many sinusoidal waves with different wavelengths and frequencies
  - Localized in space, time
  - Size designated by
    - $\Delta x$  or  $\Delta t$
  - Wave groups travel with the speed  $v_g = v_0$  of particle
- Constructing Wave Packets
  - Add waves of diff  $\lambda$ ,
  - For each wave, pick
    - Amplitude
    - Phase
  - Constructive interference over the space-time of particle
  - Destructive interference elsewhere !

Imagine Wave pulse moving along a string: its localized in time and space (unlike a pure harmonic wave)



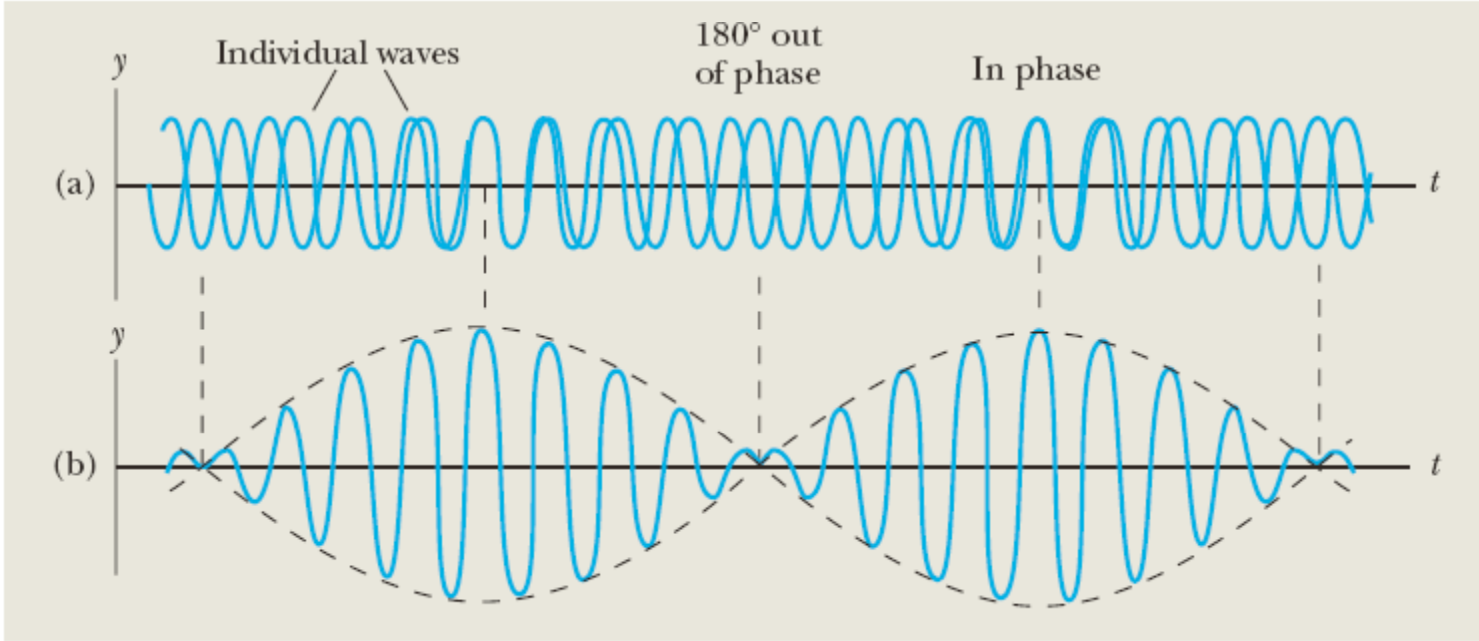
Wave packet represents particle prob





$$V_p = \frac{\omega}{k} = \lambda \cdot f$$

$$V_g = \frac{\partial \omega}{\partial k}$$





Resulting wave's "displacement"  $y = y_1 + y_2$  :

$$y = A \left[ \cos(k_1 x - \omega_1 t) + \cos(k_2 x - \omega_2 t) \right]$$

$$k = \frac{2\pi}{\lambda}$$

$$\cos(kx - \omega t)$$

Trigonometry :  $\cos A + \cos B = 2 \cos\left(\frac{A+B}{2}\right) \cos\left(\frac{A-B}{2}\right)$

$$\therefore y = 2A \left[ \left( \cos\left(\frac{k_2 - k_1}{2} x - \frac{\omega_2 - \omega_1}{2} t\right) \right) \left( \cos\left(\frac{k_2 + k_1}{2} x - \frac{\omega_2 + \omega_1}{2} t\right) \right) \right]$$

since  $k_2 \cong k_1 \cong k_{ave}$ ,  $\omega_2 \cong \omega_1 \cong \omega_{ave}$ ,  $\Delta k \ll k$ ,  $\Delta \omega \ll \omega$

$$\therefore y = 2A \left[ \cos\left(\frac{\Delta k}{2} x - \frac{\Delta \omega}{2} t\right) \cos(kx - \omega t) \right] \equiv y = A' \cos(kx - \omega t), \text{ } A' \text{ oscillates in } x, t$$

$$A' = 2A \left( \cos\left(\frac{\Delta k}{2} x - \frac{\Delta \omega}{2} t\right) \right) = \text{modulated amplitude}$$

Phase Vel  $V_p = \frac{\omega_{ave}}{k_{ave}}$

Group Vel  $V_g = \frac{\Delta \omega}{\Delta k}$

$V_g$  : Vel of envelope =  $\frac{d\omega}{dk}$

